# <u>Organic</u> LETTERS

# Enantioselective Synthesis of Allenylenol Silyl Ethers via Chiral Lithium Amide Mediated Reduction of Ynenoyl Silanes and Their Diels-Alder Reactions

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**(5)** Supporting Information

**ABSTRACT:** An enantioselective Meerwein-Ponndorf-Verley-type reduction of ynenoylsilanes by a chiral lithium amide followed by a Brook rearrangement and *anti*-mode protonation across conjugated 1,3-enynes provides allene derivatives bearing a 2-siloxyvinyl moiety in high enantioselectivity. The E/Z geometry of enol silyl ethers is controlled by the geometry of the starting enyne moiety. Thus, (E)- and (Z)enol silyl ethers are obtained from (Z)- and (E)-ynenoylsilans,



respectively. The 2-siloxyvinylallene products can participate in Diels-Alder reactions with reactive dienophiles such as PTAD, which can be achieved in a one-pot operation from ynenoylsilanes.

e have recently reported the enantioselective formation of siloxyallenes 4<sup>1</sup> bearing a vinyl substituent via Meerwein–Ponndorf–Verley-type reduction<sup>2</sup> of enynoylsilanes 1 using chiral lithium amide 2 (Scheme 1).<sup>3,4</sup> The enantioselectivity observed is explained by anti-mode S<sub>E</sub>2'-type protonation via a Brook rearrangement<sup>5</sup> in a silicate intermediate 3 or a silicate transition state.<sup>6</sup> Thus, C-Si bond cleavage and protonation in 3 occurs in a concerted fashion and in an antimode while keeping a parallel alignment between the C-Si bond and the  $\pi$  orbital of the triple bond. We became interested in examining whether the stereoelectronic and/or steric preferences could be extended to a vinylogous system 5, in which a double bond is introduced between the acylsilane moiety and the triple bond, and enol silyl ethers 7 substituted by axially chiral allenes can be formed as products. Although there are some reports on syn-type intramolecular 1,4-addition across 1,3enynes,<sup>7</sup> the stereochemical course of an anionic rearrangement across the 1,3-enyne system leading to a 1,3,4-triene system has never been explored. From a synthetic point of view,<sup>8</sup> 7 can serve as a versatile chiral building block in C-C bond formation reactions such as Diels-Alder reactions and aldol reactions by taking advantage of an axis of chirality of the allene moiety. Synthesis of allenylenol silyl ethers, however, is quite limited, and there are only a few reports in the literature even in their racemic form.<sup>9,10</sup> In this paper, we report an enantioselective synthesis of allenylenol silyl ethers using a tandem process that involves Meerwein-Ponndorf-Verley-type reduction of ynenoylsilanes by a chiral lithium amide, a Brook rearrangement in the resulting  $\alpha$ -silyl alkoxide, and protonation of a lithioallenyl carbanion.

Scheme 1. Tandem Processes for the Enantioselective Formation of Allenes



In the case using ynenoylsilanes **5** as substrates, in contrast to the case of **1**, the E/Z geometries in **5** potentially affect the E/Z geometry in the enol silvl ether moiety in the products 7,<sup>11</sup> and we therefore decided to conduct the reaction on both (*E*)- and (*Z*)-derivatives. While (*E*)-**5** was successfully prepared by Sonogashira cross-coupling<sup>12</sup> of  $\beta$ -iodoacryloyl silanes **8** with the corresponding alkynes, some modifications in the case of (*Z*)-**5** were needed; thus, the coupling of 3-iodo-1-silyl-2-propen-1-ol followed by oxidation was carried out, as shown in Scheme 2.

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Table 1. Reduction of Ynenoylsilanes 5 Using Chiral Lithium Amide 2

(Z)-10b (R = PhOCH<sub>2</sub>CH<sub>2</sub>): 79%

(Z)-5b: 79%



In initial experiments, we examined the MPV-type reduction of vnenovlsilanes (*E*)-5a with 2 in toluene at -80 °C. Although the corresponding silvl alcohol (E)-10a was obtained in an excellent enantioselectivity (er >99:1) with the absolute configuration being assigned as R on the basis of the modified Mosher method,<sup>13</sup> recovery of the starting material (7-37%)was inevitable under various conditions. The use of 2, however, generated from MeLi·LiBr and not from n-BuLi, solved this problem to provide (R,E)-10a in 85% yield (Table 1, entry 1).<sup>14,15</sup> The enantioselectivity observed can be explained by invoking a deformed six-membered transition state in which a steric interaction between the 1'-methylene group and the carbonyl-carbon substituents due to the increased double-bond character of the C-1 nitrogen bond controls the selectivity (Figure 1). That has been previously proposed for the reduction of corresponding silvlimines on the basis of the results of density functional theory (DFT) calculations.<sup>16</sup> Similar results were obtained with (Z)-**5a**,**b** as well as other (E)-substrates (Table 1).

Having established the applicability of enantioselective MPVtype reduction to ynenoylsilanes, we proceeded to examine the viability of the tandem process that involves MPV-type reduction, a Brook rearrangement in the resulting  $\alpha$ -silyl alkoxide, and the formation of an allenylenol silyl ether by



	R	0         1. 2·LiBr (1. toluene,           TBS         2. t·BuOH (           7. 7-BuOH (         THF, -80           5         5 min	2 equiv) -80 °C 1.2 equiv) ) ° to -20 °C	H OTBS	
entry	5	R	7	yield (%)	er
1	(E)- <b>5</b> a	PhCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub>	(Z)-7a	71	93:7
2	(E)- <b>5b</b>	PhOCH <sub>2</sub> CH <sub>2</sub>	(Z)-7b	78	99:1
3	(E)-5c	PhOCH <sub>2</sub>	(Z)-7c	79	90:10
4	(E)- <b>5d</b>	MeOCH <sub>2</sub>	(Z)-7d	81	94:6
5	(Z)-5a	PhCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub>	(E)-7 <b>a</b>	64	97:3
6	(Z)- <b>5b</b>	PhOCH <sub>2</sub> CH <sub>2</sub>	(E)-7 <b>b</b>	59	>99:1

Scheme 3. Determination of Absolute Configuration of (Z)-7 and (E)-7



**Figure 2.** Rotamers of  $\alpha$ -silyl alkoxide leading to (*E*)- and (*Z*)-7.

protonation across the enyne moiety. When (E)-**5a** was treated with **2** in toluene followed by addition of a THF solution of *t*-BuOH,<sup>17</sup> (*Z*)-**7a** was obtained exclusively, in sharp contrast to the result that (*Z*)-**5a** afforded only (*E*)-**7a** (Table 2). The same stereoselectivity was also observed in reactions of the other ynenoylsilanes (*E*)-**5b**-**d** and (*Z*)-**5b**. The absolute configurations of the allene derivatives (*E*)- and (*Z*)-7 were determined to be both *R* by the fact that the known allene derivative (*R*)-**11**<sup>18,19</sup> was obtained from both of the isomers as shown in Scheme 3.

The E/Z-geometry in the products is controlled by rotamers of an  $\alpha$ -silyl alkoxide, from which a Brook rearrangement occurs via a pentacoordinate-silicate transition state or intermediate to give enol silyl ethers as shown in Figure 2. This is based on the stereoelectronic requirement that the C–Si bond cleaved should be in a parallel arrangement to the  $\pi$ -bond of the double bond.

To obtain information about the origin of the strict E/Zselectivity observed, we performed DFT calculations on groundstate conformations of  $\alpha$ -silyl lithium alkoxides (E)-12a,b and (Z)-12a,b leading to (Z)- and (E)-13, respectively (B3LYP/6-311++G\*\* level; see Supporting Information).<sup>20</sup> These calculations account for the stereoselectivity observed by showing that (E)-12a and (Z)-12b leading to (Z)-13 and (E)-13 are more stable than the corresponding intermediates leading to (E)-13 and (Z)-13, respectively (Figure 3), although the origin of the different stabilities of the rotamers is not clear at present. Table 3. Diels-Alder Reactions of Allenylenol Silyl Ethers Achieved in a One-Pot Operation from Ynenoylsilanes 5



			product	(E)- <b>14</b> – <b>16</b>		(Z)- <b>14</b> – <b>16</b>	
entry	5	dienophile		yield (%)	er	yield (%)	er
$1^{a,b}$	(Z)-5a	PTAD	15a	80	99:1 (R)	16	98:2 (S)
2	(Z)-5a	NMM	14a	61	99:1 (R)	_	-
3	(Z)-5a	MA	16a	43	>99:1 (R)	_	_
$4^{a,b}$	(Z)- <b>5b</b>	PTAD	15a	62	99:1 (R)	26	99:1 (S)
5	(Z)- <b>5b</b>	NMM	14a	57	99:1 (R)	_	-
6	(Z)- <b>5b</b>	MA	16a	43	99:1 (R)	_	_
$7^a$	(E)- <b>5</b> a	PTAD	15a	51	96:4 (S)	11	97:3 (R)
8 <sup>a</sup>	(E)- <b>5b</b>	PTAD	15b	37	97:3 (S)	32	94:6 (R)
$9^a$	(E)- <b>5c</b>	PTAD	15c	35	>99:1 (S)	56	99:1 (R)
10 <sup><i>a</i></sup>	(E)- <b>5d</b>	PTAD	15d	51	96:4 (S)	30	96:4 (R)
		1					

<sup>*a*</sup>PTAD (0.5 equiv), -80 °C, 1 h. <sup>*b*</sup>TFA (0.9 equiv).



**Figure 3.** Relative energies of  $\alpha$ -silyl lithium alkoxides.

Scheme 4. Stereochemical Pathway for Protonation of  $\alpha$ -Silyl Lihium Alkoxides



The correlation between absolute configurations of the stereogenic centers of 10 and allene moieties of 7 indicates that the protonation proceeds in an anti-mode to the C-Si bond across the 1,3-envne system (Scheme 4). Our previous studies on alkynoyl silanes showed that protonation on the triple bond occurs in an anti-mode to the C-Si bond probably because of stereoelectronic and/or steric effects.<sup>3</sup> If stereoelectronic effects operate also in the 1,3-envne system, overall syn-protonation (anti, anti) products should be produced across the 1,3-enyne system. A possible explanation for the overall anti protonation observed is that the protonation is caused by t-BuOH(s) precomplexed with a lithium atom bound to an oxygen atom of a silicate or a silyl alkoxide,<sup>6b</sup> consequently from the same side as the oxygen atom and thus from the opposite side to the C-Si bond. However, a detailed understanding of the origin of the stereochemical preference for the protonation must await further mechanistic study.

To evaluate the synthetic potential of (Z)- and (E)-enol silvl ethers having an axially chiral allene moiety, we examined Diels– Alder (D–A) reactions of (R,Z)- and (R,E)-7b with *N*methylmaleimide (NMM) and *N*-phenyl-1,2,4-triazolinediones

Scheme 5. Diels-Alder Reactions of Allenylenol Silyl Ethers 7



(PTAD). The latter was selected in view of its extremely high reactivity toward a diene that allows a D–A reaction with (*Z*)-dienes under mild conditions.<sup>21</sup> The reaction of (*R*,*E*)-7**b** with NMM proceeded smoothly to give (*R*,*E*)-14**b**, while (*R*,*Z*)-7**b** resulted in recovery of the starting materials (Scheme 5). In contrast, the reactions of both (*R*,*Z*)- and (*R*,*E*)-7**b** with PTAD proceeded even at -80 °C to give (*S*,*E*)- and (*R*,*Z*)-15**b** and (*R*,*E*) and (*S*,*Z*)-14**b**, respectively. Their stereochemical assignments were made on the basis of the results of the X-ray analysis of (*R*,*Z*)-15**c** (vide infra).

These stereochemical outcomes can be rationalized by invoking that the reaction proceeds through transition states, the geometries of which are governed by a combination of steric interactions between the dienophile and the allene substituent R, and the E/Z-geometry of the enol silyl ether, as shown in Figure 4. The low E/Z selectivity in the reactions with PTAD is attributed to the extremely high reactivity of the dienophile.

We next examined whether it is possible to bypass the isolation of 1,3,4-trienol silyl ethers 7. As a result, we developed a one-pot procedure that enables the direct conversion of ynenoylsilanes 5 into 14-16 (Table 3).



Figure 4. Transition-state structures of the Diels–Alder reaction.

In conclusion, the significant points arising from the study outlined here are as follows: (1) the reduction of ynenoylsilanes via an MPV-type hydride transfer from a chiral lithium amide provides  $\alpha$ -silvl alkoxides that undergo protonation across conjugated 1,3-envnes after a Brook rearrangement to give allene derivatives bearing a 2-siloxyvinyl moiety enantioselectively; (2) the stereoselective formation of (E)- and (Z)-enol silvl ethers depending on the geometry of the starting envne moiety can be attributed to the difference between the stabilities of the rotamers leading to the Brook rearrangement; (3) the correlation between absolute configurations of the  $\alpha$ -silyl alkoxide and the allene moiety indicates that an anionic rearrangement across the 1,3-envne system followed by protonation proceeds in an antimode to the C-Si bond; (4) the 2-siloxyvinylallene products can participate in Diels-Alder reactions with reactive dienophiles, which can be achieved in a one-pot operation from ynenoylsilanes.

# ASSOCIATED CONTENT

#### **S** Supporting Information

Experimental procedure, characterization, and spectral data. This material is available free of charge via the Internet at http://pubs. acs.org.

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#### Notes

The authors declare no competing financial interest.

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